#### Modeling of Limestone Calcination Using Joule Heating

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> COMSOL Conference, Bangalore Nov 4-5, 2011

# Motivation

- Rapid heating of lime stone
  - the recent research interest evaluated by Fall (2011)
  - Savings in energy projected
- Microwave heating of Alumina
  modeled using COMSOL by Suryanarayana (2011)
- This presentation demonstrates the COMSOL capability to model joule heating of limestone with reaction.

## Model details

- Physics used
  - Joule heating
  - Transport of Diluted species
- Geometry
  - 15\*15\*1 mm limestone particle with entire surface exposed for heating
- Mesh
  - Physics controlled mesh with fine element size
- Time dependant study (0-20 min heating time)

## **Joule Heating**

- Initial condition
  - T=303.15K and V=220 Volts.
- Influx range

- 500000 (rapid heating) - 500 W/m<sup>2</sup> (slow heating).

$$\rho C_{p} \frac{\partial T}{\partial t} + \rho C_{p} \mathbf{u}_{trans} \nabla T = \nabla \cdot (k \nabla T) + Q + W_{p}$$
$$\nabla \cdot \mathbf{J} = Q_{j}$$
$$\mathbf{J} = \sigma \mathbf{E} + \frac{\partial \mathbf{D}}{\partial t} + \mathbf{J}_{e}$$
$$\mathbf{E} = -\nabla V$$

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### **Transport of Diluted Species**

 $CaCO3 = CaO + CO_2$ 

$$\frac{\partial c_i}{\partial t} + \nabla \cdot \left(-D_i \nabla c_i\right) = R_i$$

- Rate expression:  $Rj=k^*Cj$   $N_i = -D_i \nabla c_i$
- Reaction Rate Constant k= A\*exp(-E/RT)
  - Frequency factor, A=9.67\*10<sup>24</sup> hr<sup>-1</sup>
  - Activation Energy, E= 1092.947 KJ/mol
  - Convective Inward flux= 0.1 mol/m<sup>2</sup>.s
  - Mass transfer coefficient= 0.0833 m/s
  - CO2 concentration in bulk phase = 0.002 mol/m<sup>3</sup>
- Diffusion of CO<sub>2</sub>
  - DCO2 = 1.39e-8 m<sup>2</sup>/s

### Results (Heating)



# Results (CO<sub>2</sub> Concentration)

Time+1 Slice: Concentration (mol/m\*3)



 Uniform concentration of 0.014 mol/m<sup>3</sup>

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### Conclusion

- COMSOL capability demonstrated
- Useful to design the experiments

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# ACKNOWLEDGMENT

 Authors would like to thank the management of Padmasri Dr. B. V. Raju Institute of Technology for providing the Simulation laboratory facilities